

## **R.E.D.** Group of Schools

**Summer Holidays Homework Framework** 

**SESSION: 2023-24** 

 $CLASS - 12^{th}$ 

Subject: English

#### Text Book: Flamingo, Vistas and BBC Compacta

#### Syllabus Covered upto MAY END

Book Flamingo

- Chapter No.1 Chapter Name- The Last Lesson
- Chapter No.- 2 Chapter Name-Lost Spring
- Poem 1 Poem Name- My Mother at Sixty-six
- Poem 3Poem Name Keeping Quiet
- Chapter No. 3 Chapter Name- Deep Water
- Chapter No. 4 Chapter Name- The Rattrap
  - Book Vistas
- $\circ$  Chapter No. 1 Chapter Name The Third Level
- Chapter No. 4 Chapter Name- The Enemy

#### 2. List of all new concepts taughtupto MAY END (Grammar Topics)

- o Chapter No. Chapter Name- Reading Comprehension
- Chapter No. Chapter Name- Notice Writing
- Chapter No. Chapter Name- Invitations and Replies
- Chapter No. Chapter Name- Formal Letters

#### 3. Tools required for doing Homework:

- Reader Book
- o Notebook
- Scrap Book
- Resources as per activity

#### 4. Date of Submission of Homework: 3nd July, 2023

#### Category 9-12

#### 5. Instruction/Guidelines for Formative Assessment based Homework:

- Section-A-Reading and Vocabulary Homework
  - > Each student will read :

Fictional work: The Old Man and the Sea by Ernest Hemingway

Non-fiction: Restless Days, Sleepless Nights by RanjanaBharij

Write review of both the works separately using the following steps

#### > Note: - Do the following homework in scrap book

- ✓ Draw creative page as front page
- ✓ Identify and list the Main characters in the Book
- ✓ Write the summary of the story as follows :—
  - ✤ Beginning
  - ✤ Middle
  - ✤ End
- ✓ Write your favourite part of the story
- ✓ Mention anything you disliked about the book
- ✓ Book rating out of 5 and why
- $\checkmark$  If you were the author how will you end the story
- 2. Read any English newspaper once in a week and find out 5 new words from it & frame a sentence from it and present them in the same scrap book

#### **Vocabulary Homework**

Make your own dictionary.( Each student will learn 3 new words daily with meanings and write the words in dictionary )

Total 45 words should be included in your dictionary

#### • Section-B- Speaking Homework

- 1. Students will practice on one of the given topics :-
- " The limit of my language is the limit of my world"
   OR
- > Poverty and mental health

#### OR

> Mother- daughter relationship

Students will prepare speaking activity video on any one of the above topics and share with English teachers on WhatsApp group

• Section-C-Creative Writing Homework

#### **Creative Writing Homework**

**Travelogue writing:** There are many tourist attractions. They are popular for many reasons. Some places are popular for their natural beauty whereas others are for their historical and religious importance. Write the names of the places that you would like to visit in our country naturally beautiful places, places of historical and religious importance. Have you ever visited such places on your holidays? Write down the places you have visited so far. Writedetailed description of a visited place in the form of a paragraph.

Character portrait/ sketch writing of yourfavourite character from the novel"The Guide" by R.K. Narayan on A4 size sheet.

#### > Review writing: On a movie "Three idiots"

- Section-D- Learning and Pre reading Homework
  - 1. **Pre- reading**: Poem No. -4 , A Thing of Beauty

Poem No.; -5, A Roadside Stand

**Learning** : Book Flamingo

- Chapter No. 1 Chapter Name- The Last Lesson
- Chapter No.- 2 Chapter Name- Lost Spring
- Poem 1 Poem Name- My Mother at Sixty-six
- Poem 3Poem Name Keeping Quiet
- Chapter No. 3 Chapter Name- Deep Water
- Chapter No. 4 Chapter Name- The Rattrap
- o Book Vistas
- Chapter No. 1 Chapter Name The Third Level
- Chapter No. 4 Chapter Name- The Enemy
- Section-E- Project work

#### 1. Prepare a student portfolio and include the following details :-

- Personal details
- > What I understand by portfolio
- > My goals/ Aim in life for future
- > My achievements till now
- > The areas I need to work to achieve my goal

#### Following projects can be given for Grammar Topic covered in the month of April and May:-

2. 12 tenses formula with examples

#### OR

Verb project chart

#### OR

Parts of speech and application

3. Grammar flip book with all rules, tips and tricks on Clauses and Conditional sentences

OR

Draw your favorite fiction-character from the book 'Old man and the Sea' and describe it using 10 adjectives



## **R.E.D. Group of Schools**

#### **Summer Holidays Homework**

**SESSION: 2023-24** 

CLASS – 12th

Text Book: S.L. Arora, NCERT

#### Subject: PHYSICS

#### 1. Syllabus Covered upto MAY END

- Chapter No.- 01 Chapter Name- Electric charge and field
- Chapter No.- 02 Chapter Name-Electric potential and electric capacitance
- Chapter No.- 03 Chapter Name- Current and electricity
- Chapter No.- 04 Chapter Name- Magnetic effect of current

#### 2. List of all new concepts taught upto MAY END

- Force between the charges and their field
- Concept of capacitance and electric potential
- o Polarization and dielectric
- o Concept of current, Potential difference and EMF

#### 3. Formative Assessment based Homework:

- Section-A-Creative Project/ Working model/ Inquiry based project.
- Section-B-Diagram and Labeling assessment activity.
- Section-C-Experiment based activity.
- Section-D- Derivations.
- 4. Summative Assessment based Homework:
  - o Section-E- Chapter-wise Assignments

#### 5. Tools required for doing Homework:

- NCERT Text Book, S.L. Arora book
- o Notebook

- Resources as per activity

• Combination of cells and resistance

• Magnetic fields of various caring figures

o Force on a Moving charge particles in

6. Instruction/Guidelines for Formative Assessment based Homework:

#### • Section-A-Creative Project/ Working model/ Inquiry based project.

Торіс	Roll No.
Make a working project of Periscope	1 to 10
Creating Electric Current with a Magnet	11 to 20
Make Earthquake Alarm Working Model	21 to 30
Solar panel	31 to 40

#### I. (Roll No. 1 to 10) Topic: Make a working project of Periscope

• Materials Required: Two congruent pieces of mirror, cardboard or a PVC pipe, cutter, tape or glue

#### Steps to prepare:

- ▶ Use cardboard to make three hollow cuboids and arrange them in the shape of a real periscope.
- $\blacktriangleright$  Attach the mirror glasses to the opposite corners of the structure at an angle equal to 45°.
- ▶ Hold one end of the periscope on eye level and look at the distant objects easily.

#### II (Roll No. 11 to 20) Topic: Creating Electric Current with a Magnet

#### • Materials Required; Coil the large no of turns, 9V volt battery, key, Galvanomerter

#### • <u>Steps</u> to prepare:

- ▶ Wind coils using 32 gauge wire (200 turns and 400 turns). Pass a strong magnet through each of the coils.
- > Measure the amount of electricity generated by the magnet moving through the coils by using a galvanoscope.
- > Repeat the procedure by moving the magnets through the coils at different speeds to see the difference in the current generated.

#### III. Roll No.2 1 to 30) Topic: Make Earthquake Alarm Working Model

- Materials Required:- LED light (preferably red),1 kilo-ohm resistor, Wire, Buzzer, Copper • wire, Steel nut, 9v Battery, 9v Battery clip connector, On/Off switch. Cardboard
- Steps to prepare:
- > Take a 7cm long copper wire, fold it and twist it. When done, penetrate that through the

- A4 Sheets, Internet

magnetic field

perpendicular cardboard, like this.

- Take the steel nut, penetrate a copper wire through the nut and twist it. The nut would act as a pendulum in the model. Fix this steel nut in the model by taking the tail (copper wire tied to the steel nut) and putting it across the small square cardboard. After doing, fix the pendulum in its place using a glue gun. Make sure you penetrate the pendulum across the small knot using copper wire on the perpendicular cardboard piece. Fix this steel nut in the model
- Now, we need to work with the buzzer and the LED light. Firstly, take the 1-kilo ohm resistor and fix it at one end of the LED light. This can be done using a heating mechanism.
- Fix a battery to the battery connector. Using the glue gun, fix the on/off switch, the battery, buzzer and the LED light at the edge of the cardboard base to complete the model. Ensure the switch is kept on to make the model a functional earthquake alarm

#### IV. Roll No. 31 to 40) Topic: Solar panel

• <u>Materials Required;</u> ferric chloride solution, solder, solder iron, alcohol, and crystal silicon paste.

#### • <u>Steps to prepare:</u>

- > Apply crystal silicon paste over the printed circuit board and leave it to dry.
- Remove the extra paste from the printed circuit board.
- > Attach the connecting wires to form the positive and negative terminals of the solar panel.
- > Place the set-up in direct sunlight and connect a multimeter across the terminals.

#### o Section-B- Diagram and Labeling assessment activity.

- Draw well labeled diagrams of the following:
  - Equipotential surface due to 1) point charge 2) dipole 3) Line charge distribution
  - > Magnetic field due to current carrying wire
  - Magnetic field due current carrying solenoid
  - > Plots Graph electric field vs distance for point charge, line charge and dipole
  - Plots Graph of electric potential vs distance for point charge, line charge and conducting shell

#### • Section-C-Experiment based activity.

- > <u>Name of the Activity</u>: Bending Water Static Electricity Experiment
- Material Required: A sink , A comb or a balloon
- Procedure: We did this impressive science experiment in my fifth grade class, and I have remembered it all these years! This experiment can be done with a comb OR a balloon. Either one will work just fine. Turn on the faucet with a very small stream of water. The smaller, the better, but you do need the water to be running consistently and not just dripping. Then charge either the comb or the balloon by running it through your hair. Hold the comb or the balloon very close to the stream of water, but not touching it.
- > **Observations:**
- > <u>Conclusion:</u>
- > <u>Precautions:</u>

#### Section-D- Derivations

#### Learning Homework:

- Derivations
  - 1) Electric field due to dipole on its axial line and equatorial line
  - 2.) Electric field due to a Charged ring on its center and on a point on its axil line
  - 3.) Electric field due to a charged wire of infinite length using Gauss theorem
  - 4.) Electric field due to a conducting hollow shell using Gauss theorem
  - 5.) Combination of cells in series and parallel
  - 6.) Magnetic field due current carrying wire
  - 7.) Magnetic field due to current carrying circular coil on its axil line
  - 8.) Magnetic field due to a solenoid using Ampere circuital law
  - 9.) Derivation of radius, time period, frequency of a charged particle in magnetic field.
- Section-E-Revision assignment.



## **R.E.D.** Group of Schools

**Revision Assignment - 1** 

Class: 12th

Subject: Physics Ch. No.: 1

Ch. Name: Electrostatic Charges & Fields



Find -

(i) Net flux through the cylinder

(ii) Charge enclosed by the cylinder.

- (b) An early model for an atom considered it to have a positively charged point nucleus of charge Ze, surrounded by a uniform density of negative charge upto a radius R. The atom as a whole is neutral. For this model, what is the electric field at a distance r form the nucleus?
- (c) (i) Is the force acting between two point charges  $q_1$  and  $q_2$  kept at some distance in air, attractive or repulsive when : (i)  $q_1 q_2 > 0$ (ii)  $q_1 q_2 < 0$ 
  - (ii) Write down the value of absolute permittivity of free space.
  - (ii) What is the relevance of large value of K (=81) for water?
- (d) (i) A charge q is placed at the centre of a cube of side l what is the electric flux passing through two opposite faces of the cube?
  - (ii) Two concentric spherical shells of radii R and 2 R are given charges  $Q_1$  and  $Q_2$  respectively. The surface charge densities of the outer surfaces are equal. Determine the ratio  $Q_1 : Q_2$ .

#### Q.7. Assertion and reason questions:

- (a) Both Assertion and reason are true and reason is correct explanation of assertion
- (b) Both Assertion and reason are true but reason is not a correct explanation of assertion.
- (c) If assertion is true but reason is false.
- (d) If both assertion and reason are false
- Assertion : Electric dipole moment is a scalar. **i**) **Reason :**  $\vec{p} = q(2a)$
- ii) Assertion : No two electric lines of force can intersect each other. **Reason :** Because they lie far apart.
- iii) Assertion : The value of electrostatic force constant depends on nature of medium separating the charges and also on the system of units.

**Reason :** In SI,  $k = 9 \times 10^9$  Nm<sup>2</sup> C<sup>-2</sup>. In CGS system k = 1.

#### **Conceptual and Mental Ability Based Type Questions**) **Q.8**. Answer the following questions

Q1. (a) A comb run through one's hair attracts small bits of paper. Why? What happens if the hair are wet or if it is a rainy day?

(b) Ordinary rubber is an insulator. But the special rubber tyres of aircrafts are made slightly conducting. Why is this necessary?

(c) Vehicles carrying inflammable materials usually have metallic ropes touching the ground during motion. Why?

(d) A bird perches on a bare high power line, and nothing happens to the bird. A man standing on the ground touches the same line and gets a fatal shock. Why?

- $Q_2$ . If two objects repel one another, you know both carry either positive charge or negative charge. How would you determine whether these charges are positive or negative?
- **Q3.** In coulomb's law in electrostatics valid in all situations?
- Q4. Fig. shows tracks of three charged particles in a uniform electrostatic field. Give the signs of the three charges. Which particle has the highest charge to mass ratio?



- (a) Suppose two particles have identical curved trajectories. Which of the following are necessarily true?
  - (i) they have same charge
- (ii) they have same mass
- (iii) the charges have the same sign, (iv) they have the same e/m ratio.
- (b) You are given the initial velocity  $\mathcal{P}$  of a beam particle and the length of the capacitor l. What other measurement would enable one to find e/m?
- Q5. Two charges of magnitude -2 Q and + Q are located at points (a,0) and (4a,0) respectively. What is the electric flux due to charges through a sphere of radius '3 a' with its centre at the origin.



## **R.E.D.** Group of Schools

**Revision Assignment - 2** 

For recapitulation & solving the assignment the students should refer to their NCERT BOOK, MTG Part-1 (Case Study Question/Activity based Question)

Subject: Physics Ch. No.: 2

Ch. Name: Electric Potential And Capacitor

#### Instruction: Read the following passage and answer the question no. 1, 2,3,4 & 5. Case Study-1 A capacitor is an arrangement for storing a large amount of electric charge and hence electric energy in a small space. The capacitance of an isolated conductor is increased considerably by bringing near it an uncharged conductor connected to Earth. This is the principle of capacitor. Such an arrangement of the two conductors separated by a dielectric medium is said to form a capacitor or condenser depending on the shape of conductors, we get parallel plate capacitor, spherical capacitor and cylindrical capacitor. Capacity of a parallel plate capacitor is $C = A\varepsilon_0 d$ , where A is the area of plates, and d is distance between the two plates of capacitor. 0.1 When a dielectric medium of relative permittivity K is inserted between the plates of capacitor than capacitance of the capacitor will (b) decreases (a) increases. (c) remains constant (d) zero. Q.2. When condensers are joined in parallel then C= (a) $C = C_1 + C_2$ (b) $C = C_1 - C_2$ (c) $C = C_1 \times C_2$ (d) $C=1/C_1.C_2$ When condensers are joined in series then C = Q.3. (a) $C_1 C_2/C_1+C_2$ (b) $C_1 + C_2 / C_1 \cdot C_2$ (c) $C_1 + C_2$ (d) none of above What happens to capacitance of the condenser when the distance between the plates increased Q.4. (c) becomes half (d) doubles (a) increases (b)decreases Q.5. The amount of energy stored between the plates of capacitor (a) $CV^2/2$ (c) $Q^2 V/2$ (b) QV/2 (d) C2V

#### Part-2

#### Subject Specific conceptual definitions & Application based Questions

#### Q.4. Define the following terms:-

i)Electrostaticsii)Electric Dipole moment.iii)Electric polarizationiv)Electric Capacitance.

#### Q.5. Differentiate the following:-

i) Coulomb Law & Gauss Law.

- ii) Electric field line & Equipotential surface.
- iii) Electric susceptibility & Polarization density.
- iv) Electric flux & Electrostatic shielding.

#### Q.6. Application based question:-

- a) What is the flux through a cube if a Charge Q placed at (i) center (ii) at corner (iii) at center of a face
- b) What is the capacitance of a parallel plate capacitor if a dielectric slab of dielectric constant 3 and thickness x (x<d) insert between the plate of capacitor oa area of plate A and distance between the plate is d</li>

c) What is the work done to move a 2µC charge from point (5m,0) to point (0,3m) when a charge of 3µC placed at origin

#### Q.7. Assertion and reason questions:

- (a) Both Assertion and reason are true and reason is correct explanation of assertion
- (b) Both Assertion and reason are true but reason is not a correct explanation of assertion.
- (c) If assertion is true but reason is false.
- (d) If both assertion and reason are false
- Assertion : A parallel plate capacitor is connected across battery through a key. A dielectric slab of dielectric constant K is introduced between the plates. The energy which is stored becomes K times.
   Reason : The surface density of charge on the plate remains constant or unchanged.
- ii) Assertion : Two concentric charged shells are given. The potential difference between the shells depends on charge of inner shell.

Reason : Potential due to charge of outer shell remains same at every point inside the sphere.

iii) Assertion : Electric field is discontinuous across the surface of a spherical charged shell.Reason : Electric potential is continuous across the surface of a spherical charged shell.

#### **Q.8.** Conceptual and Mental Ability Based Type Questions)

#### Answer the following questions in one word or a sentence.

- Q1. Define electric flux. Write its S.I. unit.
- $Q^2$ . Why are electric field lines perpendicular at a point on an equipotential surface of a conductor?
- Q3. A ball of mass 5 g and charge 10-7 C moves from point A, whose potential is 500 V, to point B, whose potential is zero. What is the velocity of the ball at point A if, at point B, it is 25 cm per second?
- Q4. What is Gauss the theorem? Write its mathematical form .
- Q5. Write the expression for energy loss when a charged capacitor C up to potential V connect with another uncharged identical capacitor in parallel
- Q6. Find the potential energy of a system of four particles placed at the vertices square of side 'a' .Also obtain the potential at the center of the square
- Q7A capacitor of capacitance  $5.00 \ \mu\text{F}$  is charged to  $24.0 \ \text{V}$  and another capacitor of capacitance  $6.0 \ \mu\text{F}$  is charged to  $12.0 \ \text{V}$ . (a) Find the energy stored in each capacitor. (b) The positive plate of the first capacitor is now connected to the negative plate of the second and vice versa. Find the new charges on the capacitors. (c) Find the loss of electrostatic energy during the process. (d) Where does this energy go?



## **R.E.D.** Group of Schools

**Revision Assignment - 3** 

Subject: Physics Ch. No.: 3

Ch. Name: Current and electricity



- Q3. Define the term current density of a metallic conductor. Deduce the relation connecting current density (J) and the conductivity ( $\sigma$ ) of the conductor, when an electric field E, is applied to it
- Q4. State Kirchhoff s current law and voltage law

#### Q.7. Assertion and reason questions:

- (a) Both Assertion and reason are true and reason is correct explanation of assertion
- (b) Both Assertion and reason are true but reason is not a correct explanation of assertion.
- (c) If assertion is true but reason is false.
- (d) If both assertion and reason are false
- **Q.1.** Assertion : In a simple battery circuit, the point of the lowest potential is positive terminal of the battery.

**Reason :** The current flows towards the point of the higher potential, as it does in such a circuit from the negative to the positive terminal..

- Q.2. Assertion : A larger dry cell has higher emf. Reason : The emf of a dry cell is proportional to its size.
- **Q.3.** Assertion : Voltmeter is connected in parallel with the circuit. **Reason :** Resistance of a voltmeter is very large.

## Q.8. Conceptual and Mental Ability Based Type Questions)

#### Answer the following questions in one word or a sentence.

1. Two heated wires of the same dimensions are first connected in series and then it's parallel to a source

of supply. What will be the ratio of heat produced in the two cases?

- 2. The storage battery of a car has an emf of 12V If the internal resistance of the battery is  $0.4\Omega$ , what is the maximum current that can be drawn from the battery?
- 3. Potential difference Vis applied across the ends of copper wire of length 1 and diameter D. What is the effect on drift velocity of electrons if(a) V is doubled
- **4.** (a) Find the current flowing through each cell in the circuit shown in Fig. 3.181.

Also calculate the potential difference across the terminal of each cell.



(b) Find the equivalent resistance of the networks shown in Fig. 3.68 between the points A and B.



- 5. A storage battery of emf 8.0 V and internal resistance  $0.5\Omega$  is being charged by a 120 V dc supply using a series resistor of  $15.5\Omega$ . What is the terminal voltage of the battery during charging? What is the purpose of having a series resistor in the charging circuit?
- 6. A current of 2A flows through  $2\Omega$  resistor when connected across a battery. The same battery supplies a current of 0.5A when connected across a  $9\Omega$  resistor. The internal resistance of the battery is \_\_\_\_\_.
- 7. A cell of emf 'E' and internal resistance 'r' is connected across a variable load resistor R. Draw the plots of the terminal voltage V versus (i) R and (ii) the current i.
  It is found that when R = 40, the current is 14 when R is increased to 90, the current reduces to 0.5.

It is found that when  $R = 4\Omega$ , the current is 1A when R is increased to  $9\Omega$ , the current reduces to 0.5 A. Find the values of the emf E and internal resistance r

8. State Kirchhoff's rules. Use these rules to write the expressions for the currents  $I_1$ ,  $I_2$  and  $I_3$  in the circuit diagram shown.





## **R.E.D.** Group of Schools

**Revision Assignment - 4** 

**Subject: Physics** Ch. No.: 4

Ch. Name: Magnetic Effect Of Current

conductor.

For recapitulation & solving the assignment the students should refer to their NCERT BOOK,MTG Part-1

#### (Case Study Question/Activity based Question) Instruction: Read the following passage and answer the question no. 1, 2 & 3.

#### Case Study-1

A galvanometer is a device used to detect current in an electric circuit. It cannot as such be used as an ammeter to measure current in a given circuit. This is because a galvanometer is a very sensitive device. It gives a full scale deflection for a current of the order of Moreover for measuring currents. The galvanometer has to be connected in series, and it has a large resistance this will change the value of current in the circuit. To overcome these difficulties. We connect a small resistance R called shunt resistance, in parallel with the galvanometer coil, so that most of the current passes through the shunt. Now to use galvanometer as a voltmeter, it must draw a very small current, otherwise it will appreciably change the voltage which we are measuring. To ensure this a large resistance R is connected in series with the galvanometer.

- Q.1. A sensitive galvanometer like a moving coil galvanometer can be converted into an ammeter or a voltmeter by connecting a proper resistance to it. Which of the following statements is true?
  - (a) a voltmeter is connected in parallel and current through it si negligible
  - (b) an ammeter is connected in parallel and potential difference across it is small
  - (c) a voltmeter is connected in series and potential difference across it is small
  - (d) an ammeter is connected in series in a circuit and the current through it is negligible.

Q.2. By mistake a voltmeter is connected in series and an ammeter is connected in parallel with a resistance in an electrical circuit. What will happen to the instruments?

- (a) Voltmeter is damaged. (b) Ammeter is damaged (c) Both are damage.
  - (d) None of damaged

Q.3. A galvanometer coil has a resistance of 15 Ohm and gives full scale deflection for a current of 4 ma. To convert it to an ammeter of range 0 to 6 A

- (a) 10 m ohm resistance is to be connected in parallel to the galvanometer.
- (b) 10 m Ohm resistance is to be connected in series with the galvanometer.
- (c) 0.1 Ohm resistance is to be connected in parallel to the galvanometer.
- (d) 0.1 Ohm resistance is to be connected in series with the galvanometer

Q.4 Two identical galvanometers are converted into an ammeter and a milliammeter. Resistance of the shunt of milliammeter through which the current passes through will be (a) more (b) equal (c) less (d) zero

(v) A voltmeter has resistance of G ohm and range of V volt. The value of resistance used in series to convert it into a voltmeter of range **nV** volt is

	. <u>Part-2</u> Subject Specific conceptual definitions & Application based Questions					
Q.4.	Define the following terms:-					
	i) Galvanometer ii) Ammeter	iii) Voltmeter	iv) List Count			
Q.5.	Differentiate the following:-					
	i) Voltage and ammeter	ii)Ammeter and Galvanometer				
	iii) Biot Savart law and Coulomb's law	iv) Gauss law and	Ampere circuital law			
<b>Q.6</b> .	5. Application based question:-					
	i. Explain Biot-savarts law in term of					
	(i) Current density					
	(ii) Charge and its velocity.					
	ii. State Ampere's circuital law and prove	e this law for a circu	lar path around a long current carrying			

**iii**. Find the condition under which the charged particles moving with different speeds in the presence of electric and magnetic field vectors can be used to select charged particles of a particular speed.

#### Q.7. Assertion and reason questions:

- (a) Both Assertion and reason are true and reason is correct explanation of assertion
- (b) Both Assertion and reason are true but reason is not a correct explanation of assertion.
- (c) If assertion is true but reason is false.
- (d) If both assertion and reason are false
- i. Assertion (A): On changing the direction of flow of current through a straight .conductor, the direction of a magnetic field around the conductor is reversed.

**Reason** (**R**): The direction of magnetic field around a conductor can be given in accordance with left hand thumb rule.

- **ii.** Assertion (A): The magnitude of the magnetic field at a point on the axis of a current carrying solenoid is inversely proportional to the current flowing through the solenoid.
   **Reason** (R) : The magnitude of the magnetic field at a point on the axis of a current carrying solenoid is directly proportional to the number of turns per unit length of a solenoid.
- iii. Assertion (A): A compass needle is placed near a current carrying wire. The deflection of the compass needle decreases when the magnitude of an electric current in the wire is increased. Reason (R) : Strength of a magnetic field at a point near the conductor increases on increasing the current.

#### Q.8. Conceptual and Mental Ability Based Type Questions)

Answer the following questions in one word or a sentence.

- 1 Proton and a alpha particle enters in a magnetic field with same kinetic energy perpendicular to magnetic field. Find the ratio of there radius .
- 2. What is the S.I unit and dimensional formula of magnetic permeability.
- 3. In a current-carrying coil of radius R and having N turns is opened and made into a straight long wire. Then the magnetic field at a distance R would be how many times of the value of the centre of the coil?
- 4. Find the relation between the magnetic field at R/2 on-axis, and magnetic field at the centre of the coil. Here R is the radius of the coil.
- 5. Find the force per unit length on two parallel current-carrying conductors.
- 6. Find the expression for the resistance connected in parallel to convert a galvanometer to an ammeter.
- 7. Two circular coil of radius in ratio 2: 3 placed perpendicular to each other at a common center. If current in both coil is same the fine the
  - A. Ratio of magnetic field at center
  - **B.** Total magnetic field at center
  - C. Ratio of magnetic moment
- 8. Rang of a galvanometer is 0.5 A and resistance is 10  $\Omega$ . What is the value of shunt connected with galvanometer to convert it into a ammeter of range 6A?
- 9. A current carrying coil of radius R and numbers of terns N convert into a square. Find the ratio of their magnetic moment.

#### (Section-B) Lab Manual work

Links from You Tube:

Experiment-1 https://youtu.be/R j0cFkzDIY

Experiment-2.<u>https://youtu.be/JTvzP7HrMxU</u>

Experiment-3 <a href="https://youtu.be/7gllLt-BOL8">https://youtu.be/7gllLt-BOL8</a>

Experiment-4.https://youtu.be/3jIWlX4dmlI

**Experiment-1.** To determine resistivity of two / three wires by plotting a graph for potential difference versus current.

**Experiment-2**. To find resistance of a given wire / standard resistor using metre bridge.

**Experiment-3.** To determine resistance of a galvanometer by half-deflection method and to find its figure of merit.

**Experiment-4**. To convert the given galvanometer (of known resistance and figure of merit) into a voltmeter of desired range and to verify the same.

(Do this work in Practical file)

NOTE: Holiday Homework will not be accepted after the assigned date.



**R.E.D. GROUP OF SCHOOLS** SUMMER HOLIDAYS HOMEWORK (SESSION: 2023-24)

#### $CLASS - 12^{th}$

## Subject: Chemistry Text Book: NCERT book

#### 1. <u>Syllabus Covered up to MAY END</u>

- Chapter No.2 Chapter Name- Solutions
- Chapter No.3 Chapter Name- Electrochemistry.
- Chapter No.4 Chapter Name- Chemical kinetics
- Chapter No.10 Chapter Name Haloalkanes and Haloarenes
- Chapter No. 8 Chapter name: D and f block elements

#### 2. List of all new concepts taughtup to MAY END

- Colligative properties
- Concentration Terms
- Abnormal molar mass
- Electrolysis and its product
- Electrochemical cell
- Nernst equation
- ➢ Fuel cell
- Primary and secondary cell
- Preparation of Haloalkanes and Haloarenes
- > De-Hydro-halogenation reaction along with Zaitsev rule
- Nucleophilic substitution reaction and mechanism
- Physical properties of transition elements
- Transition elements
- 3. Formative Assessment based Homework:
  - Section-A-Creative Project/ Working model/ Inquiry based project.
  - Section-B-Diagram and Labeling assessment activity.
  - Section-C-Experiment based activity.
  - Section-D- Learning and Pre-reading homework.

#### 4. <u>Summative Assessment based Homework:</u>

Section-E - ChapterwiseRevisionassignment (Written, Learning & Pre-reading Homework)

#### 5. <u>Tools required for doing Homework:</u>

- 1. NCERTBook, Pradeep book
- 2. Notebook
- 3. A<sub>4</sub>Sheets
- 4. Resources as per activity

#### 6. Instruction/Guidelines for Formative Assessment based Homework:

Sr.No.	Торіс	Roll No.
1	Elevation of boiling point and depression in freezing point	1 to 10
2	A Hydrogen oxygen fuel cell	11 to 20
3	Various factors affecting rate of reaction	21 to 30
4	Nucleophilic substitution reaction of Haloalkanes and Mechanism	31 to last roll no.

#### Section-A-Creative Project/ Working model/ Inquiry based project.

1.	To stud	y the elevation of boiling point and depression in freezing point also include examples from				
	real life (Roll no. 1 to 10)					
	Material required: - A4 sheets, A Ruler, NCERT Book, A Tape and a box.					
	Steps to prepare - understand the concept and explain on A4 sheets.					
2.	To build a hydrogen oxygen fuel cell (Roll no 11 to 20)					
	<b>Steps to propage</b> - understand the concept and explain on A4 sheets					
2	<b>Steps to prepare -</b> understand the concept and explain on A4 sheets. <b>To study various factors affecting rate of reaction (Roll no 21 to 30)</b>					
3.	Material required: - A4 sheets, A Ruler, NCERT Book. A Tape and a box					
	Steps to prepare - understand the concept and explain on A4 sheets					
4.	To stud	y about nucleophilic substitution reaction of Haloalkanes and mechanism (Roll no.31 to last				
	roll no.)					
	Materia	l required: -A4 sheets, A Ruler, NCERTBook, A Tape and a box				
	Steps to	prepare - understand the concept and explain on A4 sheets.				
Sectio	on-B-	Diagram and Labeling assessment activity.				
	<b>Draw</b> y	well labelled diagrams of the following:				
	• Dra	aw a diagram of electrochemical cell of Cu and Ag				
~ .	• Dra	w the graphs of zero first and second order reaction				
Sectio	on C	Experiment based activity				
Name	of the A	<b>Activity:</b> To observe how eggshell is protected by tooth paste and learn how toothreacts				
		to acids and stain				
Mater	ial Requ	uired: egg, toothpaste, coca cola and vinegar.				
<b>Procedure</b> : Take four eggs and wash it carefully coat two plain eggs with good amount of toothpas						
		evenly. And the rest two eggs remain as it is. That means we are not coating these eggs				
		with any other material.				
		Pick one plain egg and one toothpaste coated egg and drop them in coke filled glasses				
		respectively and wait for 24 hours to see the outcomes.				
Obser	vations	Egg shell is the rich source of calcium carbonate whereas Coca-Cola is acidic in nature. When				
		the plain egg dropped in the coke, the acidic contents immediately start reacting with calcium				
		carbonate and forms stains				
		On the other hand, the eggs coated with toothpaste, when dropped in the coke and the fluoride				
		in toothpaste build a protective layer between the eggshells and acidic				
Conclu	usion:	Fluoride in toothpaste makes the eggshell stronger and protects it from reacting to acidic				
		contents of coke.				
Precau	itions:	Wash the egg carefully so that it does not break.				
<u>Sectio</u>	on-D-L	earning and Pre-reading homework.				
Learn	ing Hor	nework: Revise All N. C. E. R. T questions				
Pre-Re	eading l	Homework: Read Ch.9 coordination compounds				

#### Section-E- Revision assignment.



#### **Revision Assignment – 1**

CLASS: 12<sup>th</sup>

**Chapter Name: Solutions** 

Subject: Chemistry

Chapter No.: 2

#### Part-1

#### (Case Study Question/Activity based Question)

Instruction: Read the following passage and answer the question

#### Q.1 Case Study-1

The boiling point elevation and the freezing point depression of solutions have a number of practical applications. Ethylene glycol (CH<sub>2</sub>OH.CH<sub>2</sub>OH) is used in automobile radiators as an antifreeze because it lowers the freezing point of the coolant. The same substance also helps to prevent the radiator coolant from boiling away by elevating the boiling point. Ethylene glycol has a low vapour pressure. We can also use glycerol as an antifreeze. In order for boiling point elevation to occur, the solute must be non-volatile, but no such restriction applies to freezing point depression. For example, methanol (CH<sub>3</sub>OH), a fairly volatile liquid that boils only at 65°C is sometimes used as antifreeze in automobile radiators.

#### Answer the following questions given below: -

- (i) Out of the  $CH_3OH$  and  $C_6H_{12}O_6$ , which is a better reagent for depression in freezing point but not for elevation in boiling point?
- (ii) Will the depression in freezing point be same or different, if 0.1 moles of sugar or 0.1 moles of glucose is dissolved in 1 L of water?
- (iii) 124 g each of the two reagents glycerol and glycol are added in 5 kg water of the radiators in the two cars. Which one is better for a car? Justify your answer.

OR

If the cost of glycerol, glycol and methanol are the same, then what would be the sequence of the economy to use these compounds as antifreeze?

#### Part-2

#### (Subject Specific conceptual definitions & Application based Questions)

#### Q 2. Define the following terms:

i) Henry law ii) Osmosis iii) Colligative property iv) Van't Hoff factor

#### **Q 3.** Differentiate the following:

- i) Ideal and non-ideal solution ii) Positive and negative deviation
- iii) Minimum and maximum. Boiling azeotrope iv) Molarity and molality
- **Q 4.** Application based question:

- Calculate the osmotic pressure in Pascal's exerted by a solution prepared by dissolving 1.0 g of polymer of mass 185,000 in 450 mL of water at 37°C.
- ii) 19.5 g of CH<sub>2</sub>FCOOH is dissolved in 500 g of water. The depression in the freezing point of waterobserved is 1.0°C. Calculate the Van't Hoff factor and dissociation constant of fluoroacetic acid.
- iii) What concentration of nitrogen should be present in a glass of water at room temperature? Assume a temperature of 25°C, a total pressure of 1 atmosphere and mole fraction of nitrogen in air of 0.78. (K<sub>H</sub> for nitrogen =  $8.42 \times 10^{-17}$  M/mm Hg).
- iv) What happen when cell is placed in
  - a) 0.5% NaCl solution b) 1.5% NaCl solution

#### **Q 5.** Assertion and reason questions:

i) Assertion: The molarity of a solution in liquid state changes with temperature.

#### **Reason:** The volume of a solution changes with a change in temperature.

- (a) Both Assertion and reason are true and reason is correct explanation of assertion.
- (b) Both Assertion and reason are true but reason is not a correct explanation of assertion.
- (c) If assertion is true but reason is false.
- (d) If both assertion and reason are false.
- ii) Assertion: When NaCl is added to water a depression in the freezing point is observed.

Reason: The lowering of the vapour pressure of a solution causes depression at the freezingpoint.

- (a) Both Assertion and reason are true and reason is correct explanation of assertion.
- (b) Both Assertion and reason are true but reason is not a correct explanation of assertion.
- (c) If assertion is true but reason is false.
- (d) If both assertion and reason are false.

#### Q 6. What is van't Hoff factor for NaCl, Aluminum chloride?

- **Q 7.** What is Anoxia?
- Q 8. Why aquatic organism feelsmore comfortable in cold water?

**Revision Assignment –2** 

Chap	ter Na	me: Ele	ctrochemistry				Chapter No 3
					Part-1		
			(Case S	tudy Questio	n/Activity ba	sed Question)	
Instr	uction:	Read th	he following pas	sage and ans	wer the quest	tion	
Q.1	1 Case Study- 1						
All chemical reactions involve interaction of atoms and molecules. A large numb						ules. A large number of	
	atom	s/molect	iles are present	in a few	gram of any	chemical cor	npound varying with their
	atom	ic/molec	ular masses. To	o handle suc	h large num	ber convenien	tly, the mole concept was
	intro	duced. A	All electrochemic	al cell reaction	ons are also t	based on mole	concept. For example, a 4.0
	mola	r aqueou	s solution of Na	Cl is prepared	and 500 mL	of this solution	is electrolysed. This leads to
	the ev	volution	of chlorine gas a	t one of the el	ectrode. The a	amount of produ	icts formed can be calculated
	by us	ing mole	e concept.			I I I I	
	The	followin	g questions are i	multiple choi	ce questions.	Choose the mo	st appropriate answer:
	(i)	The to	tal number of mo	oles of chlorin	e gas evolved	l is	
		(a) 0.5	5 (	(b) 1.0	(c) 1.	5	(d) 1.9
	(ii)	If cath	ode is a Hg electro	ode, then the ma	aximum weight	of amalgam forr	ned from this solution is
		(a) 30	0g (	(b) 446 g	(c) 39	96 g	(d) 256 g
	(iii)	The to	otal charge (could	omb) required	for complete	electrolysis is	(1) 102000
		(a) 18	6000	(b) 24125	(c) 48	8296	(d) 193000
	(iv)	In the	electrolytes the	number of mo	UK Des of electro	ns involved are	
	$(\mathbf{IV})$	(a) 2	ciccuorytes, the	(b) 1	(c) 3	iis involved are	(d) 4
		(4) =		(0) 1	Part-2		
		(Sul	oject Specific co	nceptual defi	nitions & Ap	plication based	l Questions)
O 2.	Defir	ie the fo	llowing terms:	•		•	
C	(a) E	lectrolyti	ic cell	(b) Mo	olar conductiv	ity	(c) Kohlrausch law
	(d) F	uel cells		(e) Far	aday laws	5	
O 3.	Diffe	rentiate	the following:		5		
	(a)	Galva	nic and Electroly	tic cell	(b)	Primary and	secondary cell
04.	Appl	ication	based question:			j	<b>,</b>
	a)	The m	nolar conductivity	y of a 1.5 M s	olution of an	electrolyte is fo	und to be 138.9 S cm <sup>2</sup> mol <sup>-1</sup> .
	,	Calcu	late the conductiv	, vity of this sol	ution	5	
	b)	The c	onductivity of 0.	001 M acetic	acid is $4 \times 10^{10}$	0 <sup>-5</sup> S/cm. Calcu	late the dissociation constant
	- /	of ace	tic acid. if molar	conductivity	at infinite dilu	tion for acetic a	cid is 390 S cm <sup>2</sup> /mol.
	c)	Calcu	late the time to	deposit 1.27 s	g of copper at	t cathode when	a current of 2A was passed
		throug	where the solution of	CuSO <sub>4</sub> .			I I I I I I I I I I I I I I I I I I I
05.	Asse	rtion an	d reason questio	ons:			
<b>X</b> = 1	Choo	ose the c	orrect one from	. Following o	ption		
	(a)	Both A	Assertion and rea	son are true a	nd reason is co	orrect explanati	on of assertion.
	(b)	Both A	Assertion and rea	son are true b	ut reason is no	ot a correct expl	anation of assertion.
	(c)	If asse	ertion is true but i	reason is false	•		
	(d)	If both	n assertion and re	eason are false			
	Asse	rtion:	On dilution pur	r all electrolyt	es decreases o	on dilution.	
	A SCAL	rtion.	Am for weak	electrolytes s	hows a sharp	increase when	the electrolytic solution is
	1 1000		diluted.	01001101y105 5	nows a sharp		i die electrotytic solution is
	Reas	on:	For weak electr	olytes degree	of dissociatio	n increases with	n dilution of solution.



CLASS: 12<sup>th</sup>

Subject: Chemistry

**Revision Assignment –3** 

Chan	ter Name: Cł	hemical Kinetics		anject. Chemistry anter No - 4
Chap		Part-1		
		(Case Study Ouestion/Act	vity based Question)	
Instru	ction: Read	the following passage and answer th	e question	
0.1	Case Study	- 1	1	
<b>C</b> -	The half-life i.e., $[A]_t = [$ concentratio kinetics. It is finite time, infinite time <b>The followi</b> <b>O1.For the</b>	e of a reaction is the time required for $[A]/2$ . For first order reaction, $t_{1/2} = 0$ on. Figure shows that typical variation may be noted that though the major p but the reaction will never cease as e. ng questions are multiple choice que half-life period of a first order rest	or the concentration of react $0.693/k$ this means $t_{1/2}$ if a of concentration of react portion of the first order k the concentration of react estion. Choose the most ap action. which one of the	tant to decrease by half, s independent of initial ant exhibiting first order inetics may be over in a ant will be zero only at <b>propriate answer:</b> following statements is
	generally fa	alse?	,	8
	(a) it is inde	pendent of initial concentration	(b)It is independent of ter	mperature.
	(c) it decreas	ses with the introduction of a catalyst.	(d) None of these.	-
	Q2.The rate	e constant for a first order reaction	is 7.0 x 10 <sup>-4</sup> s <sup>-1</sup> . If initial co	ncentration of reactant
	is 0.080 M,	what is the half-life of reaction?		
	(a) 990 s	(b) 79.2 s	(c)12375 s	(d) 10.10 x 10 <sup>-4</sup> s
	Q3. A react	tion's rate constant is k= $3.28 \times 10^{-4}$ s	<sup>-1</sup> . Determine the reaction	's order.
	(a) First orde	er (b) Second order	(c) Third order	(d) Fourth order
Q.2	Define the f	following terms: -		
-	a) Activation	n Energy	c) Pseudo first order reac	tion
	b) Half-life		d) Rate constant	
Q3	Differentiat	te the following:		
	a) Order and	d Molecularity	b) zero and first order rea	action
Q.4	Application	based question: -		
	a) Expl	an effect of temperature and catalyst	on rate of reaction with graj	phs
	b) A II	rst-order reaction is 50 percent com	plete in 30 minutes. Calc	ulate the time taken for
	com]	pretion of 87.5 percent of the reaction.	$\mathbf{v} = \mathbf{v}$	V V V shance if the
	C) HOW	entration of the substance X is doubled	d and that of V is halved?	$\rightarrow \Lambda_2$ i change if the
05	Assertion a	nd reason questions.		
Q.J	Choose the	correct one from Following ontion		
	(a)If both A	ssertion and Reason are correct and	l Reason is correct explan	ation of Assertion.
	(b)If both A	Assertion and Reason are correct but	R is not the explanation	of Assertion.
	(c)If Assert	ion is correct Reason is wrong.	<b>F</b>	
	(d)If Assert	tion is wrong Reason is correct.		
1.	Assertion:	Precipitation of silver chloride oc	curs instantaneously by mi	xingaqueous solution of
		silver nitrate and sodium chloride.		
	Reason:	Ionic reactions occur very fast		
2.	Assertion:	Order and molecularity are same.		
	Reason:	Order is determined experimentall	y and molecularity is the s	sum of thestoichiometric
		coefficient of rate determining elem	nentary step.	



CLASS: 12<sup>th</sup>

Subject: Chemistry

## **Revision Assignment –4**

CLA	SS: 12 <sup>th</sup>	1				Subject: Chemistry		
Chap	ter Nai	me- Hal	oalkanes	and Haloarenes		Chapter No 10		
Part	1 Case	based q	uestion					
	Nucle	eophilic	substitutio	on reactions are of two	types; substitution nucleo	philic bimolecular $(S_N 2)$ and		
	substitution nucleophilic unimolecular (S <sub>N</sub> 1) depending on molecules taking part in determin							
	rate of reaction. Reactivity of alkyl halide towards S <sub>N</sub> 1 and S <sub>N</sub> 2 reactions depends on various factors							
	such as steric hindrance, stability of intermediate or transition state and polarity of solvent. $S_N 2$							
	reaction mechanism is favoured mostly by primary alkyl halide or transition state and polarity of							
	solvent, $S_N 2$ reaction mechanism is favoured mostly by primary alkyl halide then secondary and then							
	tertiar	ry. This o	order is re	versed in case of S <sub>N</sub> 1 r	eactions.	-		
	Answer the following questions given below:							
	(i)	Which	n of the fo	llowing is most reacti	ve towards nucleophilic s	substitution reaction?		
	(-)	(a)C <sub>6</sub> F	4.Cl	(b)CH <sub>2</sub> =CHCl (c	$c)ClCH_2CH=CH_2$ (d) C	H₃CH=CHCl		
	(ii)	Isonro	nvl chlor	ide undergoes hydrol	vsis hv			
	(11)	$(a)S_{N1}$	mechanis	sm	$(b)S_{N}2$ mechanism			
		$(c)S_{N}1$	and Sy2	mechanism	(d)neither Syl nor S	v2 mechanism		
	(iii)		nost react	ive nucleonhile among	the following is	N2 meenamism		
	(111)		105116act	(b)C.H.O	(c)(CH <sub>2</sub> )-CHO <sup>-</sup>	$(d)(CH_{2})_{2}CO^{2}$		
02	Dofin	(a)CII	llowing	$(0)C_{6}I15O$	(C)(CII3)2CIIO	(u)(CII3)3CO		
Q2		mbidant	nualaanh	ile				
	$\begin{array}{c} a \\ b \\ \end{array}$		ation					
	(0) $(0)$	uitz iea	citon	<b>n</b>				
02	С) ГІ <b>D:ff</b>							
QS		rentiate	the follow	ving				
04			SINZ Mech					
Q4	Application based question: -							
	a)	HOW V	VIII YOU DI	ing about the following	g conversions?			
		l. 	Ethane to	b bromo-etnene				
	1 \	II.	But-I-en	e to but-2-ene				
	6)	A hyd	lrocarbon	$C_5H_{10}$ does not react	with chlorine in dark bu	t gives a single monochloro		
	,	compo	bund $C_5H_9$	CI in bright sunlight. Ic	dentify the hydrocarbon.			
	c)	Predic	t all the a	lkenes that would be for	ormed by dehydrohalogen	ation of the following halides		
		with so	odium eth	oxide in ethanol and id	entify the major alkene:			
		III.	I-Bromo	-1-methylcyclohexane				
	•	iv.	2-Chlord	-2-methylbutane				
	d)	Out of	$C_6H_5CH_2$	CI and $C_6H_5CHClC_6H$	5, which is more easily hyc	irolysed by aqueous KOH.		
	e)	p-aicn	lorobenze	ne nas nigner m.p. a	ind lesser solubility than	those of o-and m-isomers.		
05	Asser	tion and	os. d reason (	mestions				
Q5	Choo	se the co	orrect on	e from. Following opti	ion			
	(a)	If bot	h Assertio	on and Reason are cor	rect and Reason is corre	ct explanation of Assertion.		
	<b>(b)</b>	If botl	h Assertio	on and Reason are cor	rect but R is not the expl	anation of Assertion.		
	(c)	If Ass	ertion is o	correct Reason is wron	ng.			
	( <b>d</b> )	If Ass	ertion is v	vrong Reason is corre	ect.			
1	Asser	tion	: E	Sutan-2-ol is optically a	ctive.			
	Reasc	on tion	: It	s mirror image is non-	superimposable on it.			
2.	Asser	uon	: 1 a	ryl halides.	group facilitates nucleop	onnic substitution reaction in		
	Reaso	on	: Т	he intermediate carban	ion is stabilized due to pre	sence of the nitro- group.		



#### **Revision Assignment –5**

CLASS: 12<sup>th</sup>

#### Chapter Name- D and F block elements

Subject: Chemistry Chapter No.- 8

#### 1. Read the passage given below and answer the following questions:

The transition elements have incompletely filled d-subshells in their ground state or in any of their oxidation states. The transition elements occupy position in between s- and p-blocks in groups 3-12 of the Periodic table. Starting from fourth period, transition elements consists of four complete series: Sc to Zn, Y to Cd and La, Hf to Hg and Ac, Rf to Cn. In general, the electronic configuration of outer orbitals of these elements is  $(n - 1) d^{1-10} ns^{1-2}$ . The electronic configurations of outer orbitals of Zn, Cd, Hg and Cn are represented by the general formula  $(n - 1)d^{10} n^2$ . All the transition elements have typical metallic properties such as high tensile strength, ductility, malleability. Except mercury, which is liquid at room temperature, other transition elements have typical metallic structures. The transition metals and their compounds also exhibit catalytic property and paramagnetic behaviour. Transition metal also forms alloys. An alloy is a blend of metals prepared by mixing the components. Alloys may be homogeneous solid solutions in which the atoms of one metal are distributed randomly among the atoms of the other

## (i) Which of the following characteristics of transition metals is associated with higher catalytic activity?

	(a) High enthalpy of atomisation	(b) Variable oxidation states
	(c) Paramagnetic behaviour	(d) Colour of hydrated ions
(ii)	Transition elements form alloys	easily because they have
	(a) same atomic number	(b) same electronic configuration
	(c) nearly same atomic size	(d) same oxidation states.
(iii)	The electronic configuration of t	tantalum (Ta) is
	(a) $[Xe]4f^{0}5d^{1}6s^{2}$	(b)[Xe) $4f^{14}5d^26s^2$
	(c) $[Xe]4f^{14}5d^36s^2$	$(d)[Xe]4f^{14}5d^{4}6s^{2}$
(iv)	Which one of the following our	ter orbital configurations may exhibit the largest number of
	oxidation states?	
	(a) $3d^54s^1$	(b) $3d^54s^2$
	$(c)3d^24s^2$	$(d)3d^{3}4s^{2}$
( <b>v</b> )	Which one of the following out	er orbital configurations may exhibit the smallest number of
	oxidation states?	
	(a) $3d^54s^1$	(b) $3d^54s^2$
	(c) $3d^24s^2$	(d) $3d^34s^2$
Q2	Define the following	
	a) Transition elements	b) Paramagnetic substance

#### Q3 Application based question: -

#### Explain giving reason:

- (a) Transition metals and many of their compounds show paramagnetic behaviour.
- (b) The enthalpies of atomisation of the transition metals are high.
- (C) Describe briefly the following physico-chemical properties of transition metals
  - (i) Metallic character (ii) Complex formation.
- (d) Copper (I) is diamagnetic whereas copper (II) is paramagnetic. Explain.
- (e) The colour of a solution of  $K_2Cr_2O_7$  depends on pH of the solution. Transition elements show variable valencies. What is the reason for it and how is this variation different from that shown by the p-Block elements?
- (f) What happens when
- (i)  $K_2Cr_2O_7$  reacts with acidified solution of KI
- (ii) SO<sub>2</sub> is passed through an acidic solution of potassium dichromate
- (iii)  $K_2Cr_2O_7$  reacts with sodium chloride in the presence of conc.  $H_2SO_4$ .

#### Q4 Assertion and reason questions:

#### Choose the correct one from. Following option

- (a) If both Assertion and Reason are correct and Reason is correct explanation of Assertion.
- (b) If both Assertion and Reason are correct but R is not the explanation of Assertion.
- (c) If Assertion is correct Reason is wrong.

#### (d) If Assertion is wrong Reason is correct.

- **1** Assertion : Zinc does not show characteristic properties of transition metals.
  - **Reason** : In Zn outermost shell is completely filled.
- **2. Assertion** : The highest oxidation state of chromium in its compounds is +6.
  - **Reason** : Chromium atom has only six electrons in ns and (n 1) d-orbitals

#### (Section-B)

#### Lab Manual work

#### Complete the following experiment in practical file

- 1. To determine the strength and molarity of KMnO<sub>4</sub> solution by titration against mohr salt
- 2. To determine the percentage purity of  $KMnO_4$  solution by titration against mohr salt
- 3. To determine the strength and molarity of KMnO<sub>4</sub> solution by titration against oxalic acid
- 4. To determine the percentage purity of KMnO<sub>4</sub> solution by titration against oxalic acid



### **R.E.D.** Group of Schools

**Summer Holidays Homework** 

Session: 2023-24

**Text Book: NCERT Text Book** 

#### Class –12th

#### 1. Syllabus Covered upto MAY END

- o Chapter No.- 1..... Chapter Name:- Relation and Function
- o Chapter No.- 2..... Chapter Name:- I.T.F
- o Chapter No.- 3..... Chapter Name:- Matrices
- Chapter No.- 4..... Chapter Name:- Determinants
- o Chapter No.- 5..... Chapter Name:- Differentiation

#### 2. List of all new concepts taught upto MAY END

- o Number of reflexive relation, Symmetric relation
- $\circ~$  Number of one-one function, onto function
- o Properties of I.T.F
- o Formula to find area of triangle using determinants
- $\circ~$  Derivative of a function with respect to another function

#### 3. Formative Assessment based Homework:

- o Section-A-Creative Project/ Model
- Section-B- Problem solving activities.
- Section-C- Mental Maths problems.
- Section-D- Lab practicals

#### Section-A-Creative Project/ Models

Торіс	Roll .No
1. Equivalence Relation	1 to 8
2. Formulas of I.T.F	9 to 16
3. Matrices ( types, notation and properties)	17 to 24
4. Continuity and differentiability	25 to 32
5. Number of solution of system of linear equations by real life examples	33 to 40
Make a project files according to the given topics	

#### • Section-B- Problem solving activity.

Solve the following real life based problems.

- Show that the function  $f(x) = x^3 2x^2 + 2x$ ,  $x \in Q$  is increasing on Q.
- The daily profit, P, of an oil refinery is given by

 $P = 8x - 0.02x^2$ ,

• where *x* is the number of barrels of oil refined. How many barrels will give maximum profit and what is the maximum profit?

#### o Section-C- Activities related to Mental Maths.

#### Mental Maths problems:

- Q1. Look at this series: 2, 1, (1/2), (1/4), ... What number should come next?
- Q2. Look at this series: 22, 21, 23, 22, 24, 23, ... What number should come next?
- Q3. Complete the series 1,6,13,22,33,...
- Q4. Complete the Series 34,45,56,67.....
- Q5. a,b,c are in H.P. if a,b,c are in .....

Q6. If E is the universal set and  $A = B \cup C$ , then the set E - (E - (E - (E - (E - A)))) is same as the set

Q7. Which one of the following is not a prime number?

(a) 31 (b) 61 (c) 71 (d) 91

Q8. What least number must be added to 1056, so that the sum is completely divisible by 23?

(a)2 (b) 3 (c) 18 (d) 21

Q9. The sum of first five prime numbers is:

(a)11 (b) 18 (c) 26 (d) 28

Q10. The smallest 3 digit prime number is:

(a)101 (b) 103 (c) 109 (d) 113

#### Section-D-Lab Practicals .

- Make the following lab activities in lab manual.
- 1. A relation R is symmetric but Neither Reflexice Not Transitive.
- 2. A Function which is One-One but not Onto
- 3. Continuity at a Point

\*

> Section-E- Revision assignments (Chapter wise assignments).



#### **R.E.D.** Group of Schools Revision Assignment -1

Class: 12 <sup>th</sup> Ch. No.: Relation and function		Subject: Maths Ch. Name: 1
For recapitulation & solving the as	ssignment the students should refer to t	heir NCERT text book of Maths
	<u>Part-1</u>	
<u>Multiple c</u>	hoice Questions (only one option is a	<u>correct)</u>
1. If R is a relation on the set $A =$	$\{1, 2, 3\}$ given by R = $\{(1, 1), (2, 2),$	(3, 3), then R is
(a) reflexive	(b) symmetric only	
(c) transitive only	(d) equivalence relation	
2. If a function $f : R \to R$ is defined	ed by $f(x) = x^2 + 1$ , then per-images of	f 17 and $-3$ respectively, are
(a) $\phi$ , {4, -4}	(b) $\{3, -3\}, \phi$	
$(c){4,-4}, \phi$	$(d){4, -4}, {2, -2}$	
3. If a function $f: C \to C$ is defined	ed $f(x) = 3x^2 - 1$ , where C is the set of	complex numbers, then the pre-
images of $(-28)$ are		
(a) 3, – 3	(b) 3 <i>i</i> , –3 <i>I</i>	
(c) $3i$ only	(d) $-3i$ only	
4. If a function $f : [2, \infty) \rightarrow R$ is	defined by $f(x) = x^2 - 4x + 5$ , then the r	cange of $f$ is
(a) R	(b) [1, ∞)	
(c) [4, ∞)	(d) [5, ∞)	
5. If function $f : R \to R$ is defined by $f(x) = \sin x$ and function $g : R \to R$ is defined by $g(x) = x^2$ , then		
(fog) (x) is		
(a) $x^2 \sin x$	(b) $(\sin x)^2$	
(c) $\sin x^2$	(d) $(\sin x)/x^2$	
	Part – II	

#### (Integer Type Questions)

- 6. The number of bijective functions from set A to itself when A contains 106 elements is
- 7. The maximum number of equivalence relations on the set  $A = \{1, 2, 3\}$  are
- 8. Let  $E = \{1, 2, 3, 4\}$  and  $F = \{1, 2\}$  Then, the number of onto functions from E to F is
- 9. If the set A contains 5 elements and the set B contains 6 elements, then the number of one-one and onto mappings from A to B is

#### <u> Part – III</u>

#### (Application Based Questions)

#### Answer the following questions by applying acquired knowledge, facts, techniques and rules

- 10. State the reason for the relation R in the set  $\{1, 2, 3\}$  given by  $R = \{(1, 2), (2, 1)\}$  not to be transitive.
- 11. Let  $f: R \{-4/3\} \rightarrow R$  be a function given by f(x) = 4x/(3x+4). Show that f is invertible with  $f^{-1}(x) = 4x/(4-3x)$
- 12.  $F(x) = x + 7andg(x) = x 7, x \in R$ , find (fog) (7).

#### <u>Part – IV</u> (Reason and Numerical Based Questions)

- 13. Let N be the set of natural numbers and relation R on N be defined by  $R = \{(x, y) : x, y \in N, x + y + 10\}$ . Determine whether the above relation is reflexive, symmetric and transitive.
- 14. Let  $f: X \to Y$  be a function. Define a relation R in X given by  $R = \{(a,b): f(a) = f(b)\}$ . Show that R is an equivalence relation.

#### <u>Part – V</u> (Case Study Based Questions)

15. A relation R in a set A is called Reflexive: if  $(a, a) \in R$ , for every  $a \in A$  Symmetric: if  $(a_1, a_2) \in R$ ,

implies that  $(a_2, a_1) \in R$ , for all  $a_1, a_2 \in R$  Transitive: if  $(a_1, a_2) \in R$ ,  $a_2, a_3 \in R$  implies that  $a_1, a_3 \in R$  for all  $a_1, a_2, a_3 \in A$ .

Based on the above information, answer the following questions

- 1. Relations R in the set A= $\{1,2,3,...,13,14\}$  defined as R= $\{(x,y) : 3x-y=0\}$ 
  - (a) Reflexive but not symmetric nor transitive
  - (b) Neither Reflexive nor symmetric nor transitive
  - (c) Reflexive and symmetric but not transitive
  - (d) Equivalence relation
- 2. Relation R in the set Z of all integers defined as  $R = \{(x,y) : x y \text{ is an integer}\}$ 
  - (a) Equivalence Relation
  - (b) Reflexive but neither symmetric nor transitive
  - (c) Reflexive and transitive but not symmetric
  - (d) Symmetric but not transitive and reflexive
- 3. Relation R in the set A of human beings in a town  $R{(x,y) : x \text{ is the father of } y}$ 
  - (a) Transitive but not reflexive nor symmetric
  - (b) Reflexive but not symmetric nor transitive
  - (c) Neither reflexive nor symmetric nor transitive
  - (d) Equivalence Relation
- 4. Relation R in the set A of human beings in a town  $R = \{(x,y) : x \text{ is wife of } y\}$ 
  - (a) Neither reflexive nor symmetric but transitive
  - (b) Reflexive not symmetric not transitive
  - (c) Equivalence Relation
  - (d) Symmetric but not reflexive not transitive

#### <u>Part – VI</u>

#### (Analysis Based Question)

#### Answer the following question by organizing and integrating the information.

16. Consider  $f: R_+ \to [-9, \infty)$  given by  $f(x) = 5x^2 + 6x - 9$ . Prove that f is invertible with

$$f^{-1}(y) = \left(\frac{\sqrt{54+5y}-3}{5}\right)$$

**Learning Homework:** Learn all definitions & formulas from Page No. 1.1 to 1.5 of Elements Book. **Pre-Reading Homework:** Read Page no1.1 to 1.5 of Elements Book and understand their meaning

**Revision Assignment-2** 

**Ch. Name: Inverse Trigonometric Function** For recapitulation & solving the assignment the students should refer to their NCERT text book of Maths Part-1 Multiple choice Questions (only one option is correct) 1. The equation  $2\cos^{-1} x + \sin^{-1} x = \frac{11\pi}{6}$  has (a) No solution (b) Only one solution (c) Two solutions (d) Three solutions 2. If  $\tan^{-1} x + \tan^{-1} y + \tan^{-1} z = \pi$ , then x + y + z is equal to (b) 0 (c) 1 (d) 2*xyz* (a) *xyz* 3. If  $\tan^{-1}\frac{a+x}{a} + \tan^{-1}\frac{a-x}{a} = \frac{\pi}{6}$ , then  $x^2 =$ (b)  $\sqrt{3}a$  (c)  $2\sqrt{3}a^2$  (d) None of these (a)  $2\sqrt{3}a$ 4. Find the value of  $\cot^{-1}\frac{xy+1}{x-y} + \cot^{-1}\frac{yz+1}{y-z} + \cot^{-1}\frac{zx+1}{z-x} =$ (a) 0(b) 1 (c)  $\cot^{-1} x + \cot^{-1} y + \cot^{-1} z$ (d) None of these 5. If  $\alpha = \tan^{-1}\left(\frac{\sqrt{3}x}{2y-x}\right), \beta = \tan^{-1}\left(\frac{2x-y}{\sqrt{3}y}\right)$ , then  $\alpha - \beta =$ (a)  $\frac{\pi}{6}$  (b)  $\frac{\pi}{3}$  (c)  $\frac{\pi}{2}$  (d)  $-\frac{\pi}{3}$ 

#### <u>Part – II</u>

#### (Integer Type Questions)

6. Find the positive integral solution of the equation  $\tan^{-1} x + \cos^{-1} \frac{y}{\sqrt{1+y^2}} = \sin^{-1} \frac{3}{\sqrt{10}}$ 

7. Find the number of real solution of the equation  $\sqrt{1 + \cos 2x} = \sqrt{2} \sin^{-1}(\sin x), -\pi \le x \le \pi$ 

8. If  $\sin^{-1}(x^2 - 7x + 12) = n\pi$ ,  $\forall n \in I$ , then find x.

9. If  $6\sin^{-1}(x^2 - 6x + 8.5) = \pi$ , then find *x*.



Class: 12<sup>th</sup>

Subject: Maths Ch. No.: 2

#### <u>Part – III</u>

#### (Application Based Questions)

Answer the following questions by applying acquired knowledge, facts, techniques and rules

10. Prove that 
$$\cot^{-1}\left(\frac{1+\cos x}{\sin x}\right) = \frac{x}{2}$$
.  
11. Prove that  $\tan^{-1}\left(\frac{\sqrt{1+x}-\sqrt{1-x}}{\sqrt{1+x}+\sqrt{1-x}}\right) = \frac{\pi}{4} - \frac{1}{2}\cos^{-1}x$ .  
12. Prove that  $\tan^{-1}\frac{1}{2} + \tan^{-1}\frac{2}{11} = \tan^{-1}\frac{3}{4}$ .  
**Part - IV**  
(Reason and Numerical Based Questions)  
13. Evaluate  $:\cos\left[\sin^{-1}\frac{1}{4} + \sec^{-1}\frac{4}{3}\right]$ .  
14. Prove that  $\tan\left[\frac{1}{2}\sin^{-1}\frac{2x}{1+x^2} + \frac{1}{2}\cos^{-1}\frac{1-x^2}{1+x^2}\right] = \frac{2x}{1-x^2}$ .  
**Part - V**  
(Case Study Based Questions)  
15. If  $f(x) = \tan^{-1}x(\frac{2x}{1-x^2})$   
Based on the above information, answer the following questions  
i.  $f(x)$  is equal to  
(a)  $\tan^{-1}x$  (b)  $2\tan^{-1}x$   
(c)  $\cot^{-1}x$  (d)  $\sin^{-1}(\frac{1-x^2}{1+x^2})$   
ii.  $f(x)$  is equal to  
(a)  $\pi/2 + \cot^{-1}x$  (d)  $\tan^{-1}(\frac{1-x^2}{1+x^2})$   
ii. Derivative of  $f(x)$  is equal to  
(a)  $\frac{1}{2}\frac{1}{1+x^2}$  (b)  $\frac{1}{1+x^2}$  (c)  $\frac{-1}{1+x^2}$  (d)  $\frac{-2}{1+x^2}$   
iv.  $f(x)$  is equal to  
(a)  $\cos^{-1}(\frac{1+x^2}{1-x^2})$  (b)  $\cos^{-1}(\frac{1-x^2}{2x})$   
(c)  $\cos^{-1}(\frac{1+x^2}{1-x^2})$  (d)  $\cos^{-1}(\frac{1-x^2}{2x})$   
(c)  $\cos^{-1}(\frac{1+x^2}{1+x^2})$  (d)  $\cos^{-1}(\frac{1+x^2}{2x})$   
**Part - VI**  
(Analysis Based Question)

Answer the following question by organizing and integrating the information.

16. (i) If 
$$\cos^{-1}\frac{x}{a} + \cos^{-1}\frac{y}{b} = \alpha$$
 prove that  $\frac{x^2}{a^2} - 2\frac{xy}{ab}\cos\alpha + \frac{y^2}{b^2} = \sin^2\alpha$ .  
(ii)  $(\tan^{-1}x)^2 + (\cot^{-1}x)^2 = \frac{5\pi^2}{8}$ , then find *x*.

**Revision Assignment-3** 

For recapitulation & solving the assignment the students should refer to their NCERT text book of Maths

Class: 12<sup>th</sup> Ch. Name: Matrix and Determinants Subject: Maths Ch. No.: 3

Part-1 Multiple choice Questions (only one option is correct) 1. If  $A = \begin{bmatrix} \cos \alpha & \sin \alpha \\ -\sin \alpha & \cos \alpha \end{bmatrix}$ , then  $A^2 =$ (a)  $\begin{bmatrix} \cos 2\alpha & \sin 2\alpha \\ \sin 2\alpha & \cos 2\alpha \end{bmatrix}$ (b)  $\begin{bmatrix} \cos 2\alpha & -\sin 2\alpha \\ \sin 2\alpha & \cos 2\alpha \end{bmatrix}$ (c)  $\begin{bmatrix} \cos 2\alpha & \sin 2\alpha \\ -\sin 2\alpha & \cos 2\alpha \end{bmatrix}$ (d)  $\begin{bmatrix} -\cos 2\alpha & \sin 2\alpha \\ -\sin 2\alpha & -\cos 2\alpha \end{bmatrix}$ 2. If  $A = \begin{bmatrix} 1 & 2 & 3 \\ -2 & 3 & -1 \\ 3 & 1 & 2 \end{bmatrix}$  and *I* is a unit matrix of  $3^{rd}$  order, then  $(A^2 + 9I)$  equals (a) 2A (b) 4A (c) 6A (d) None of these 3. If  $A = \begin{bmatrix} 1 & \tan \theta/2 \\ -\tan \theta/2 & 1 \end{bmatrix}$  and AB = I, then B =(a)  $\cos^2 \frac{\theta}{2} A$ (b)  $\cos^2\frac{\theta}{2}.A^T$ (d) None of these (c)  $\cos^2\frac{\theta}{2}.I$  $A = \begin{bmatrix} 1 & \sin \theta & 1 \\ -\sin \theta & 1 & \sin \theta \\ -1 & -\sin \theta & 1 \end{bmatrix}, \text{ where } 0 \le \theta \le 2\pi. \text{ Then,}$ (a) Det (A) = 0 (b)  $Det(A) \in (2, \infty)$ (c)  $Det(A) \in (2,4)$  (d)  $Det(A) \in [2,4]$ 5. If  $\begin{vmatrix} 2x & 5 \\ 8 & x \end{vmatrix} = \begin{vmatrix} 6 & 5 \\ 8 & 3 \end{vmatrix}$ , then x =(b)  $\pm 3$  (c)  $\pm 6$  (d) 6 (a) 3 <u>Part – II</u> (Integer Type Questions) 6. Find the trace of the matrix  $A = \begin{bmatrix} 1 & -5 & 7 \\ 0 & 7 & 9 \\ 11 & 8 & 9 \end{bmatrix}$ . 7. Find the number of possible matrices of order  $3 \times 3$  with each entry 2 or 0.  $5^2$   $5^3$   $5^4$ 8. Find the value of given determinant  $\begin{bmatrix} 5^3 & 5^4 & 5^5 \\ 5^4 & 5^5 & 5^6 \end{bmatrix}$ 9. Evaluate:  $\begin{vmatrix} \log_3 512 & \log_4 3 \\ \log_3 8 & \log_4 9 \end{vmatrix} \times \begin{vmatrix} \log_2 3 & \log_8 3 \\ \log_3 4 & \log_3 4 \end{vmatrix}$ 

Part – III (Application Based Questions) Answer the following questions by applying acquired knowledge, facts, techniques and rules 3  $A = \begin{bmatrix} \mathbf{2} \\ \mathbf{5} \end{bmatrix}$  $-2^{j}$ , write  $A^{-1}$  in terms of A. 11. Let A be a square matrix of order 3 x 3. Write the value of |2A|, where |A|=4. 1 1 12. Find the maximum value of  $1 + \sin \theta$ 1 1  $1 + \cos \theta$ Part – IV (Reason and Numerical Based Ouestions) 13. If  $\begin{bmatrix} xy & 4 \\ z+6 & x+y \end{bmatrix} = \begin{bmatrix} 8 & \omega \\ 0 & 6 \end{bmatrix}$ , find the values of x, y, z and  $\omega$ . 14. Verify that  $A = \begin{vmatrix} 2 & 3 \\ 1 & 2 \end{vmatrix}$  satisfies the equation  $A^3 - 4A^2 + A = O$ . Part – V (Case Study Based Questions) 15. If A=[aij]  $m \times n$  matrix, then the matrix obtained by interchanging the rows and columns of A is called the Transpose of A. A square matrix A=[aij] is said to be symmetric if  $A^{T}=A$  for all possible values of i and j. A square matrix A=[aij] is said to be skew symmetric if  $A^{T} = -A$  for all possible values of i and i. Based on the above information answer the following questions: 5 (i) Find the transpose of matrix:  $\begin{bmatrix} \frac{1}{2} \\ -1 \end{bmatrix}$ (b)  $\begin{vmatrix} \frac{-1}{5} \\ 2 \\ -1 \end{vmatrix}$  $\begin{array}{c} \left\lfloor \frac{1}{2} \\ 5 \\ \end{array} \right\rfloor \\ \left( c \right) \left[ 5 \quad \frac{1}{2} \quad -1 \right] \end{array}$ (d) None of these (ii) Using transpose properties,  $(ABC)^{T}$  is equal to: (a)CTBTA<sup>T</sup> (b)  $ATBTC^{T}$ (b) (c)  $ATCTB^{T}$ (d)  $BTATC^{T}$ (iii) For any square matrix A with real number entries: (a)  $A+A^{T}$  is symmetric and  $A-A^{T}$  is a skew-symmetric matrix. (b)  $A+A^{T}$  is skew-symmetric and  $A-A^{T}$  is a symmetric matrix. (c)  $A+A^{T}$  is a symmetric as well as a skew-symmetric matrix. (d)  $A-A^{T}$  is a skew-symmetric as well as a symmetric matrix. (iv) Any square matrix can be expressed as: (a) Difference of a symmetric and a skew-symmetric matrix. (b) Sum of two symmetric matrices. (c) Sum of a symmetric and a skew-symmetric matrix. (d) Sum of two skew-symmetric matrices. Part – VI (Analysis Based Question) Answer the following question by organizing and integrating the information. b+c c+a a+b16. If a, b and c are real numbers and  $\Delta = \begin{vmatrix} c+a & a+b \\ b+c \end{vmatrix} = 0$  then show that either a + b + c = 0 or a = b = c. a+b b+c c+a

**Revision Assignment-4** 





continuous function.

**Revision Assignment-5** 

Class: 12<sup>th</sup> Ch. Name: Differentiation

Subject: Maths Ch. No.: 5

For recapitulation & solving the assignment the students should refer to their NCERT text book of Maths Part-1 Multiple choice Questions (only one option is correct)  $\frac{d}{dx}\log\tan\left(\frac{\pi}{4}+\frac{x}{2}\right) =$ (c)  $\sec x$  (d)  $-\sec x$ (b) – cosec x(a) cosec x2. If  $x = t^2$ ,  $y = t^3$ , then  $\frac{dy}{dx} =$ (c)3/2t (a)3/2(d)3t/2 (b)3/4t3. The derivative of  $\sec^{-1}\left(\frac{1}{2x^2+1}\right)$  with respect to  $\sqrt{1+3x}$  at x = 1/3. (a) does not exist (b) 0(d) 1/3 $(c) \frac{1}{2}$ 4. If sin(x + y) = log(x + y), then  $\frac{dy}{dx} =$ (a) 2 (b) -2 (c) 1 (d) -1 5. Let  $U = \sin^{-1} \left( \frac{2x}{1+x^2} \right)$  and  $V = \tan^{-1} \left( \frac{2x}{1-x^2} \right)$ , then  $\frac{dU}{dV} =$ (b) x (c)  $\frac{1-x^2}{1+x^2}$  (d) 1 (a)  $\frac{1}{2}$ Part – II (Integer Type Questions) 6. Given  $y = \sin 2x$ , then find dy/dx at x = 37. If  $x^y \cdot y^x = 16$ , then find the value of dy/dx at (2, 2) 8. If  $y = (1 + x)(1 + x^2)(1 + x^4)....(1 + x^{2n})$ , then find the value of dy/dx at x = 09. Find the value of c for which the Mean value theorem holds for the function  $f(x) = \log_e x$  on the interval [1, 3]. Part – III (Application Based Questions) Answer the following questions by applying acquired knowledge, facts, techniques and rules 10. If  $x = a\left(\cos t + \log \tan \frac{t}{2}\right)$  and  $y = a\sin t$ , find the value of  $\frac{dy}{dt} at t = \frac{\pi}{4}$ . 11. Differentiate  $\tan^{-1}\left(\frac{1+\cos x}{\sin x}\right)$  with respect to x. 12. If  $x = a(2\theta - \sin\theta)$  and  $y = a(1 - \cos 2\theta)$  find  $\frac{dy}{dx}$  when  $\theta = \frac{\pi}{3}$ . <u> Part – IV</u> (Reason and Numerical Based Questions) 13. If  $y = \cot^{-1}\left(\frac{\sqrt{1+\sin x} + \sqrt{1-\sin x}}{\sqrt{1+\sin x} - \sqrt{1-\sin x}}\right)$ ; prove that  $\frac{dy}{dx}$  is independent of x. 14. If  $y = \log\left(\sqrt{x} + \frac{1}{\sqrt{x}}\right)$ , prove that  $\frac{dy}{dx} = \frac{x-1}{2x(x+1)}$ .

#### <u> Part – V</u>

#### (Case Study Based Questions)

15. The derivative of f at x= c is defined by:  $f'(x) = \lim_{h \to 0} \frac{f(c+h) = f(c)}{h}$ . A function is said to be differentiable at a point c if left hand derivative at x=c is equal to the right hand derivative at x=c.Similarly, a function is said to be differentiable in an interval (a,b), if it is differentiable at every point of (a,b).

Based on the above information, answer the following questions:

(i) Derivative of  $f(x) = \cos(\sqrt{x})$  is:

- (a)  $-\sin(\sqrt{x})$  (b)  $-\sin(\sqrt{x})/2\sqrt{x}$
- (c)  $\sin(\sqrt{x})$  (d)  $1/2\sin(\sqrt{x})$

(ii) If y=sint, x=acost then dy/dx is:

(a)  $\cos t$  (b)  $-\tan t$  (c)  $-\cot t$  (d)  $\sin t$ 

(iii) f(x)=|x| is:

(a) Differentiable at all points  $x \in R$ 

(b) Differentiable at all points  $x \in \mathbb{R} - \{0\}$ 

(c) Not Differentiable at x=1

(d) None of these

(iv) Derivative of function  $f(x)=\sin(x^2)$  is:

- (a)  $2\cos(x^2)$  (b)  $2x\cos(x^2)$
- (c)  $2x^2 \sin(x)$  (d)  $2\cos(x)$

#### <u>Part – VI</u>

#### (Analysis Based Question)

#### Answer the following question by organizing and integrating the information.

16. (i) If  $y = x^{\sin x} + (\sin x)^{\cos x}$ , find  $\frac{dy}{dx}$ . (ii)  $y = x^{\cos x} + (\cos x)^{\sin x}$ , find  $\frac{dy}{dx}$ .



#### Summer Holidays Homework (Session: 2023-24)

#### **Class-XII**

#### **Subject- Phy. Education**

#### Text Book: Phy. Education (Ratnasagar)

#### "Learning is the only thing the mind never exhausts, never fears and never regrets"

The long awaited Summer Vacation is here. It allows you to rejuvenate and catch up on hobbies and other interests that are best pursued with time. However, it is important to strike balance between leisure and learning. To enhance your learning, we have planned Home Assignment, to keep your skill sharp and concepts clear. It will help you develop vital skills such as independent research.

# As it is well recognized that R.E.D. Sr. Sec. School not only focuses on academics but lay equal importance on Co-Scholastic Competencies. The school also desires you to adhere to the following guidelines for a fulfilling break:

- Read newspapers and magazines to gather knowledge of different processes used in physical education related industries and technological applications.
- Develop process-skills and problem- solving abilities and application of physical education
- The Holidays home work must be done in a very neat and presentable manner. Questions must be done in the given sequence.
- The child will be assessed for the presentation, neatness, completion of all the given questions and timely submission.
- Make sure that all syllabus done till June must be revised thoroughly as you begin your Pre-Mid from July.
- The revision assignments will help you prepare for Pre-Mid in the month of July.
- For the project work, wherever it is mentioned, strictly adhere to the instructions.
- Holiday Home work is a part of Internal Assessment and Practical Exam Assessment.

#### Syllabus Covered:

Chapter-1: Management of Sporting Events

Chapter-2: Children and women in sports

Chapter-3: Yoga as preventive measure for Lifestyle Disease

Chapter-4: Physical education and sports for (CWSN) Divyang

Daily Time to be spent for doing H.W (Hours per day): 45 minutes.

#### **Tools Required**

Physical Education Book, Notebook, Pen, Chart, Pencil.

#### Home Work Assessment by Parents

Parents will match the quantity of homework given with the index of student's homework notebook and certify it on the last page.

Date of Submission: 03-07-2023

#### **Checkpoints for the Assessment of Home Work:**

Sr. No.	<b>Checkpoints for Assessments of Home work</b>	Remarks
1	Index	Updated /Not updated
2	Quality of Handwriting presentation	Good /Avg./ Poor
3	Quality of creative writing	Good /Avg./ Poor
4	Quality of Project work	Good /Avg./ Poor
5	Quality of Reading & Comprehension skills	Good /Avg./ Poor
6	Whether Learning work done or not	Done/ Not done
7	Whether H.W is fully Completed	Yes/ No
8	Whether H.W is partly Pending	Yes/ No

R.E.D. GROUP OF SCHOOLS					
		Holiday Homewor	rk Assignment – I		
Clas Ch	ss: XII Nomo: Monogo	mont of Sporting Evonts		Subject: Phy. Edu. Ch. No. : 1	
Con	Name: Manage				
1	All questions are	compulsory			
1. 2.	The assignment of	compulsion y.	ompulsory.		
3.	Question 1-8 car	ry 1 mark each and are Multiple Cho	oice Questions.		
4.	Questions 9 carr	y 2 marks each and shall not exceed 4	40-60 words.		
5. 6	Questions 10-11 c	arry 3 marks each and shall not exce	eed 80 -100 words.		
0. 7.	Questions 12 car Questions 13 - 14	ry 4 marks each and shall not exceed 4 carry 5 marks each and shall not ex	ceed 150-200 words.		
8.	Summer Holiday	work should be written in a fair Not	te book.		
		Sectio	on - A		
Q.1	Which committe	e is responsible for selecting the off	ficials involved in a sports	event?	
	(a) Reception Co	ommittee	(b) Boarding and lodging	g Committee	
~ <b>^</b>	(c) Committee for	or Officials	(d) Equipment Commit	tee	
<b>Q</b> .2	In, Gi ability	eater number of matches are played	by the teams, Every team	s gets a rank according to his	
	(a) Knockout	(b) League	(c) Seeding	(d) Special seeding	
Q.3	Assertion (A):	In Knock out tournament minimu	m number of officials is rea	uired in organizing	
		tournaments.			
	Reason (R):	In Knock out tournament the team the tournament.	n once defeat immediately of	eliminated from	
	Codes:				
	(a) Both A and	R are true, but R is not the correct e	explanation of A		
	(b) Both A and	R are true, and R is the correct expl	anation of A		
	(c) A is true, bu	it R is false			
0.4	(d) A is false, b	ut R is true			
Q.4	(a) Pup for fun	lowing is not a category of run?	(b) Pup for unity		
	(a) Kun Ior Iun (b) Kun Ior unity (c) Run for identity				
0.5	Which Sports co	mpetition is organized within the sc	chool its self.		
(a) Intramural (b) Extramural (c) Interstate (d) None of these					
Q.6	What is the basic	c function of management is?		ζ,	
	(a) Controlling	(b) Budgeting	(c) Planning	(d) Organizing	
Q.7	What is the anot	her name of League tournaments?			
	(a) Knock out To	ournaments	(b) Round Robin Tourna	ments	
	(c) Combination	Tournaments	(d) None of these		
Q.8	How many byes	will be given if there are 22 teams?	(a) 11		
	(a) 10	(D) 12	(C) 11	(d) 6	
00	Differentiat	e between Intramural and Extramur	ol <u>D – II</u>		
Q.9	Differentiat		ai.		
0.10	) Write a shee	Section Section Management	$\frac{1}{2}$		
Q.10	J Write a shol	the have of 12 Teams on the basis of the	ll. alvant fintuna mith fann taar	and and a	
Q.1.	I Draw a fixti	LIE OF 15 TEAMS ON THE DASIS OF KNO	$\mathbf{n} - \mathbf{D}$	ns seeded.	
Q.12	2 Write a shore	rt Note On Specific Sports Program Sectio	$\underline{\mathbf{F}}_{\mathbf{m}}$ me: Run For Unity, Run For $\mathbf{F}_{\mathbf{m}}$	or Fun, Sports Days.	
Q.13	3 List down th	ne Advantages and Disadvantages o	f Knock out and League To	ournaments.	
Q.14	4 Explain the	Various committees and their Respo	onsibilities.		

	<b>R.E.D. GROUP OF SCHOOLS</b>			
	Holiday Homework Assignment – 2			
Class: XII		Subject: Phy. Edu.		
Ch. Na	me: Children and women in sports	Ch. No.: 2		
Genera	Il Instruction:			
1. All	questions are compulsory.			
$\begin{array}{ccc} 2. & 1n \\ 2 & 0n \end{array}$	e assignment consists of 14 questions and all are compulsory.			
3. Qu 4. Ou	estions 9 carry 2 marks each and shall not exceed 40-60 words.			
5. Qu	estions10-11 carry 3 marks each and shall not exceed 80 -100 words.			
6. Qu	estions 12 carry 4 marks each and shall not exceed 100-150 words.			
7. Qu	estions 13 - 14 carry 5 marks each and shall not exceed 150-200 words.			
<b>ð.</b> 5u	Section – A			
0 1	Which one of the following is not a type of spinal curvature deformity?			
Q. 1.	(a) Kyphosis (b) Round shoulder (c) Scoliosis	(d) Lordosis		
Q. 2.	The inward curvature of the spine is commonly known as			
	(a) Kyphosis (b) Round shoulder (c) Scoliosis	(d) Lordosis		
Q. 3.	What is the leading risk factor for global deaths according to WHO?			
	(a) Lack of Physical activity (b) Poor Nutritic	on		
0.4	(c) Smoking (d) Excess Physical Action in the first second	ical activity		
Q. 4.	Which exercise should be taken into consideration in order to improve that	foot deformity?		
	(a) Avoid arrying heavy weight			
	(c) Don't force babies to start walking at a very early stage			
	(d) All of the above			
Q. 5.	Which deformity is just opposite to knock knee position?			
	(a) Round shoulder (b) Flat foot (c) Knock knees	d) Scoliosis		
Q. 6.	Amenorrhoea is a condition of			
	(a) Irregular menstrual cycle (b) Absence of r	nenstrual cycle		
07	(c) Normal menstrual cycle (d) None of these (d) None of these (d) None of these (d) None of these (d) None of the corrective every set of the	ie No		
Q. 7.	Assertion (A): Sconosis is much tougher to treat with concentre exercise Reason (R): One must perform abdominal strengthening exercise	e.		
	Codes'			
	(a) Both A and R is true, but R is not the correct explanation of A.			
1	(b) Both A and R is true, and R is the correct explanation of A.			
	(c) A is true, but R is false.			
	(d) A is false, but R is true.			
Q. 8.	A healthy normal menstrual period is called	a (d) None of these		
	(a) Menopause (c) $20$ days cycle Section – B	e (d) None of these		
0.9	Differentiate between anorexia nervosa and bulimia nervosa			
<b>X</b> • >•	Section – C			
0 10	Discuss female athlete's triad in detail			
$\bigcirc 11$	Discuss about Special Consideration in details			
Q. 11.	Section – D			
0.12	Discuss menetrual disfunction and its types			
Q. 12.	<b>Section</b> $-\mathbf{F}$			
$\underline{\mathbf{DCUUH} - \mathbf{E}}$ O 13 What is good posture? Explain its causes and Advantages of Correct posture				
Q. 13.	Fundain the Common postural deformities in details	16.		
Q. 14.	Explain the Common postular deformities in defails.			

	<b>R.E.D. GROUP OF SCHOOLS</b>		
	Holiday Homework Assignment – 3		
Class: XII			Subject: Phy. Edu.
Ch. Name: Yoga as Preventive Measure for Lifestyle Disease			Ch. No.: 3
Gene:	ral Instruction: Il questions are compulsory		
1. A 2. T	he assignment consists of 14 questions and all are co	mpulsory.	
3. Q	uestion 1-8 carry 1 mark each and are Multiple Cho	ice Questions.	
4. Q	puestions 9 carry 2 marks each and shall not exceed 4	10-60 words.	
5. Q 6. Q	puestions 12 carry 4 marks each and shall not exceed	100-150 words.	
7. Q	uestions 13 - 14 carry 5 marks each and shall not ex	ceed 150-200 words.	
8. S	ummer Holiday work should be written in a fair Not	e book.	
0.1.	<u>Sectio</u> Neeti along with her father was regular at Distr	<u>n – A</u> ict Park in early morning. S	She realized that most of the
<b>X</b> , 1,	children are obese. She along with her few cl	assmates wanted to help the	ose children. She discussed
	with her physical education teacher and the	Principal of the school. S	School decided to organize
	awareness rally for the neighbourhood.		
	A. Obesity causes		
	(a) Underweight	(b) Diabetes	
	(c) Back pain	(d) Both b & c	
	B. Which of the following Asana (posture) is a	not used for curing obesity.	
	(a) Ardhmatsyendrasana	(b) Vajrasana	
(c) Parvatasana (d) Trikonasana			
	C. Choose the Asana which is used for curing	obesity	
(a) Sukhasana (b) Shavas		(b) Shavasana	
	(c) Vajrasana	(d) Shalabhasana	
Q. 2.	Which asana is also known as Diamond Pose.		
	(a) Tadasana	(b) Vahrasana	
	(c) Shalavhasana	(d) Shavasana	
Q. 3.	Which of the following factors does not cause o	f obesity?	
	(a) Genetic	(b) Frequency of eating	
	(c) Psychological factor	(d) none of these	
Q. 4.	Which asana is helpful in maintaing normal blo	od pressure?	
	(a) Shavasana	(b) Padmasana	
0.5	(c) Shalavasana	(d) Vakrasana	
Q. J.	(a) Lungs	(b) Eves	
	(c) Joints	(d) Stomach	

Q. 6.	Ramya's mother was suffering from leg and back pain. She has undergone many treatments, but still,			
	she is suffering from the problem. At last she decided to meeta yoga instructor and discussed her			
	problem. He prescribed some asana.			
	A. Which of the following is suitable for her mother's problem?			
	(a) Vakrasana	(b) Shalabhasana		
	(c) Tadasana	(d) Ardhmatsyendrasana		
	B. Which of the following asana helps to p	Which of the following asana helps to prevent obesity?		
	(a) Vajrasana	(b) Chakrasana		
	(c) Sukhasana	(d) Matsyasana		
	C. Which among this is a sitting asana?			
	(a) Ardhmatsyendrasana	(b) Trikonasana		
	(c) Ardha Chakarasana	(d) Hastasana		
Q. 7.	Symptoms such as coughing, wheezing,	shortness of breath and/or chest tightness are caused in		
	patients suffering from:			
	(a) Diabetes	(b) Obesity		
	(c) Asthma	(d) Back pain		
Q. 8.	In which disease, blood glucose, or blood su	igar levels are too high.		
	(a) Obesity	(b) Diabetes		
	(c) Hypertension	(d) Sinusitis		
	Se	<u>ction – B</u>		
Q. 9.	Explain about yoga as preventive measure i	n Details.		
	Se	<u>ction – C</u>		
Q. 10.	Write the three benefits of sheetali Pranayar	na.		
Q. 11.	Discuss about Special Consideration in details.			
	<u>Se</u>	<u>ction – D</u>		
Q. 12.	What is yogmudra? Write any three benefits	s of Yogmudra in brief.		
	<u>Se</u>	ection –E		
Q. 13.	What is hypertension? Draw a stick diagram	ns of two any asana recommended to control Obesity and		
	explain their procedure.			
Q. 14.	Explain the Common postural deformities in	n details.		



## Holiday Homework Assignment – 4

Class: XII			Subject: Phy. Edu.
Ch. Na	me: Physical education and sports for (CWSN) D	ivyang	Ch. No.: 4
Genera	l Instruction:		
1. All	questions are compulsory.		
2. Th	e assignment consists of 14 questions and all are comp	ulsory.	
3. Qu	estion 1-8 carry 1 mark each and are Multiple Choice	Questions.	
4. Qu	estions 9 carry 2 marks each and shall not exceed 40-6	0 words.	
5. Qu	estions10-11 carry 3 marks each and shall not exceed a	80 -100 words.	
6. Qu	estions 12 carry 4 marks each and shall not exceed 100	)-150 words.	
7. Qu	estions 13 - 14 carry 5 marks each and shall not exceed	d 150-200 words.	
8. Su	mmer Holiday work should be written in a fair Note b	ook.	
	Section –	A	
Q. 1.	In which year the Paralympics word was used offic	ially?	
	(a) 1948 (b) 1976	(c) 1960	(d) 1988
Q. 2.	When was the Deaflympic start?		
0.0	(a) 1927 (b) 1924	(c) 1957	(d) 1960
Q. 3.	What is the most important, while dealing with CW	SN?	(d) All of these
0.4	(a) Time (b) Patience What is the motto of paralympic games?	(c) Sympathy	(d) All of these
Q. <del>4</del> .	(a) Sprit of motion (b) Sprit in motion	(c) Motion of sprit	(d) Motion in sprit
Q. 5.	Which city host first deaflympic?		(d) filodon in spire
	(a) Sofia (b) Malmo	(c) Paris	(d) London
Q. 6.	WADA is related to		
_	(a) Fitness (b) Doping	(c) Health	(d) Life skill
Q. 7.	How many countery participate in 2015 deaflympic	es?	
	(a) 95 (b) 96	(c) 97	(d) 98
Q. 8.	The Term Paralympic was officially used in which	summer paralympic gar	nes.
	(a) 1984 (b) 1988	(c) 1992 D	(d) 2000
0 9	What was the thought behind starting special Olym	<u>D</u> nics	
Q. J.		C	
Q. 10.	Explain about special Olympic Games in details.	<u> </u>	
Q. 11.	Discuss about Paralympics in details.		
	<u>Section –</u>	D	
Q. 12.	Discuss about special Olympic rules in briefs.		
0.15	<u>Section –</u>	<u>E</u>	
Q. 13.	Explain the advantages of physical activities for CV	WSN.	able for CWCN
Q. 14.	what is disability? Explain the strategies to make p	mysical Activities assess	sable for C w SIN.